

Grade & Course: Physical Science 9-12	Topic: Energy	Duration: 4 weeks
Teachers: MHS Physical Science PLC Teachers	Unit 4 Energy	
Georgia Standards and Content:		
<p>SPS4. Obtain, evaluate, and communicate information to explain the changes in nuclear structure as a result of fission, fusion and radioactive decay.</p> <p>A. Develop a model that illustrates how the nucleus changes as a result of fission and fusion.</p> <p>B. Use mathematics and computational thinking to explain the process of half-life as it relates to radioactive decay. (Clarification statement: Limited to calculations that include whole half-lives.)</p>		
<p>SPS7. Obtain, evaluate, and communicate information to explain transformations and flow of energy within a system.</p> <p>a. Construct explanations for energy transformations within a system. (Clarification statement: Types of energy to be addressed include chemical, mechanical, electromagnetic, light, sound, thermal, electrical, and nuclear.)</p> <p>b. Plan and carry out investigations to describe how molecular motion relates to thermal energy changes in terms of conduction, convection, and radiation.</p> <p>c. Analyze and interpret specific heat data to justify the selection of a material for a practical application (e.g., insulators and cooking vessels).</p> <p>d. Analyze and interpret data to explain the flow of energy during phase changes using heating/cooling curves.</p>		
<p>Topics to Cover:</p> <ul style="list-style-type: none">• Types of Energy (chemical, mechanical, electromagnetic, light, sound, thermal, electrical, and nuclear)• Nuclear - Fission/Fusion - Radioactive Decay and Half-life• Molecular Motion - conduction, convection, and radiation• Specific Heat - insulation• Phase changes - heating and cooling curves		
<p>Lesson Content:</p> <p>Definitions of Energy: Energy is a quantitative property of a system that depends on the motion and interactions of matter and radiation within that system. • That there is a single quantity called energy is due to the fact that a system’s total energy is conserved, even as, within the system, energy is continually transferred from one object to another and between its various possible forms. • At the macroscopic scale, energy manifests itself in multiple ways, such as in motion, sound, light, and thermal energy.</p> <p>Conservation of Energy and Energy Transfer • Conservation of energy means that the total change of energy in any system is always equal to the total energy transferred into or out of the system. • Energy cannot be created or destroyed, but it can be transported from one place to another and transferred between systems.</p> <p>Energy in Chemical Processes and Everyday Life • Although energy cannot be destroyed, it can be converted to less useful forms—for example, to thermal energy in the surrounding environment. • Energy is also stored in the electric fields between charged particles and the magnetic fields between magnets, and it changes when these objects are moved relative to one another. • The term “heat” as used in everyday language refers both to thermal energy (the motion of atoms or molecules within a substance) and energy transfers by convection, conduction, and radiation (particularly infrared and light). • In science, heat is used only for this second meaning; it refers to energy transferred when two objects or systems are at different temperatures. • Temperature is a measure of the average kinetic energy of particles of matter. • The relationship between the temperature and the total energy of a system depends on the types, states, and amounts of matter present. • The amount of energy transfer needed to change the temperature of a matter sample by a given amount depends on the nature of the matter, the size of the sample, and the environment. Energy is transferred out of hotter regions or objects and into colder ones by the processes of conduction, convection, and radiation.</p>		
Narrative / Background Information		
Prior Student Knowledge: (REFLECTION – PRIOR TO TEACHING THE UNIT)		
Units 1-4 Atomic Structure and Nuclear Reactions, Periodic Table, Chemical Bonding and Chemical Reactions, Atomic & Molecular motion, and Forces and Motion laid the foundation for completion of this unit.		

The students rising to 9th grade in Fall 2021 may never have seen the 8th grade science classroom at all.

[Link to GSE 8th Grade Science](#)

These students have been exposed to the 8th Science GSE that lays the foundation for the high school Physical Science standards.

For this Unit and the ones that follow:

Students will need a basic knowledge of algebra and the basics of atomic and molecular structure

Year-Long Anchoring Phenomena: (LEARNING PROCESS) Operation of cars

Unit Phenomena (LEARNING PROCESS)

Candles can be used to power a toy car. <https://youtu.be/1LqL6lR8XI>

The candle-powered car is an application of the Seebeck Effect (Thermoelectric Effect). This effect is the result of thermal energy conversion directly into electricity. This phenomenon can be used to show energy conversion from heat to electricity to the kinetic energy of the car. A more detailed explanation will be required related to electron response to temperature differences in different materials. The Seebeck circuit used in this candle-powered car can also be connected to a voltmeter and used as a temperature sensing thermocouple.

Amazon Link to purchase car (if requested):

<https://www.amazon.com/Exergia-4032066066009-Candle-Car-Kit/dp/B00GA095MI>

Nuclear Fission & Fusion - Nuclear Applications

There is a great deal of energy stored in the nucleus of an atom that can be harnessed for electrical power production but the use of nuclear power does come with risks.

[Chernobyl Video](#)

Inquiry Statement: Scientific and technical innovations allow us to observe, investigate, and analyze the movement and transfer of energy between systems in order to design products with desired features.

Global Context/Exploration: Scientific and technological innovation

Science & Engineering Practices:

SEP:

- Develop and Use Models
- Construct explanations
- Plan and carry out investigations
- Analyze and interpret data

ATL:

- Make inferences and draw conclusions
- Collect, record, and verify data
- Use mathematics and computational thinking

Thinking (or critical thinking): Draw justifiable conclusions based on processing, interpreting and evaluating data gained from scientific investigations.

Communication (or interaction): Use appropriate scientific terminology, data tables and graphs to make the meaning of your findings clear to an audience of your peers.

Disciplinary Core Ideas:

(KNOWLEDGE & SKILLS)

Types of Interactions
The strong and weak nuclear interactions are important inside atomic nuclei—for example, they determine the patterns of which nuclear isotopes are stable and what kind of decays occur for unstable ones.

Definitions of Energy
Conservation of Energy and Energy Transfer
Energy in Chemical Processes

Crosscutting Concepts:

(KNOWLEDGE & SKILLS)

**Energy and Matter
Systems and Systems Models**

Key and Related Concepts:

Systems (Key)

Related:
Energy
Movement

Possible Preconceptions/Misconceptions: (REFLECTION – PRIOR TO TEACHING THE UNIT)**Common Misconceptions about Energy**

1. Energy is truly lost in many energy transformations.
2. There is no relationship between matter and energy.
3. If energy is conserved, why are we running out of it?
4. Energy can be changed completely from one form to another (no energy losses).
5. Things “use up” energy.
6. Energy is confined to some particular origin, such as what we get from food or what the electric company sells.
7. An object at rest has no energy.
8. The only type of potential energy is gravitational.
9. Gravitational potential energy depends only on the height of an object.
10. Doubling the speed of a moving object doubles the kinetic energy.
11. Energy is a “thing.” This is a fuzzy notion, probably because of the way we talk about newton-meters or joules. It is difficult to imagine an “amount” of an abstraction.
12. The terms “energy” and “force” are interchangeable.
13. From the non-scientific point of view, “work” is synonymous with “labor.” It is hard to convince someone that more “work” is probably being done playing football for one hour than studying an hour for a quiz.

Key Vocabulary: (KNOWLEDGE & SKILLS)

Types of Energy and Conservation of Energy	Nuclear Energy	Thermal Energy	Optional: Energy Sources and the Environment
energy work force system chemical energy mechanical energy electromagnetic energy light sound thermal energy electrical nuclear energy kinetic energy potential energy elastic potential energy chemical potential energy gravitational potential energy law of conservation of energy projectile motion power friction	fission fusion radioactive decay half-life isotopes	temperature thermal energy heat specific heat heat transfer average kinetic energy conduction convection convection currents radiation solar radiation thermal insulator solar collector thermodynamics heat engine heat pump internal combustion engine heating curve cooling curve phase changes	fossil fuel combustion reactions petroleum fractional distillation nonrenewable resources natural gas coal renewable resources photovoltaic cell hydroelectricity wind energy geothermal energy biomass energy transformations environmental impacts population carrying capacity pollutant hazardous waste photochemical smog acid precipitation agriculture deforestation urban development

Inquiry Questions:

Factual - What is the energy associated with motion?

What is potential energy?

What type of energy is stored in the nucleus of the atom?

	<p>on.com%2Fview%3Fid%3D5e2ac4c1-260d-4a67-abae-8b0d03213b63</p> <p>Calculating Heat Energy https://app.discoveryeducation.com/learn/player/4d15aa4b-bc9f-4750-9b78-cb88aea67ffd</p> <p>Students will be able to explain the concept of heat capacity and how to figure out the amount of energy transferred between substances through taking notes, performing practice questions, and participating in demonstrations.</p> <p>Energy Transformations Conduction, Convection, and Radiation</p> <p>GaVS - Heat Transfer http://cms.gavirtualschool.org/Shared/Science/PhysicalScience15/Energy_Shared/PhysicalScience_Energy_Shared6.html</p> <p>Specific Heat https://teaching.betterlesson.com/lesson/635246/specific-heat</p> <p>Explore through Explain Part II: PPT & Guided Notes (Balloon Phenomenon can be substituted in place of soda bottle/blow torch). (Stop before going into calorimetry).</p> <p>http://cms.gavirtualschool.org/Shared/Science/PhysicalScience15/Energy_Shared/PhysicalScience_Energy_Shared7.html</p>		
Week 2	<p>Nuclear Energy Nuclear Reactions: Fission & Fusion w/ Chernobyl Phenomenon</p> <p>Direct Instruction - fusion, fission, spontaneous radioactive decay</p> <p>Penny Half-Life introduction</p>	<p>Guided Practice - Chernobyl half-life calculation</p> <p>Penny Half-life lab</p>	<p>Half-life of group isotope calculations (C-13, O-18, U-235, U-238, Bi-213)</p> <p>Students need to calculate half-lives and present/write on how soon they will be inert (4 half-lives)</p> <p>Differentiate - Students briefly research the use of their isotope they calculated and present/turn in audio recording</p>

			Penny half-life lab graphs https://drive.google.com/file/d/1WqAf4zQEr89Pls6LYQ5sNhM-io_AHUr/view?usp=sharing
Week 3:	<p>Analyze and interpret specific heat data to justify the selection of a material for a practical application (e.g., insulators and cooking vessels). Too Hot to Handle Virtual Lab https://app.discoveryeducation.com/learn/player/6932e175-5707-4469-bbdc-97cd6957882b</p> <p>Analyze and interpret data to explain the flow of energy during phase changes using heating/cooling curves.</p> <p>CK12 Exploration Phase Change Simulation https://interactives.ck12.org/simulations/chemistry/phases-of-matter/app/index.html?screen=sandbox</p> <p>Students will review and practice the previous knowledge gained.</p>	<p>Heating and Cooling Curves Heating and Cooling Curves Pre-Assessment (What do students recall from States of Matter?)</p> <p>https://www.ck12.org/c/chemistry/heating-and-cooling-curves/asmtpractice/Heating-and-Cooling-Curves-Practice/?collectionCreatorID=3&conceptCollectionHandle=chemistry--heating-and-cooling-curves&collectionHandle=chemistry#</p>	<p>Too Hot to Handle Virtual Lab Completion</p> <p>CFA</p> <p>CSA</p>
Differentiation Strategies	<ul style="list-style-type: none">● Student Choice● Shared interest centers● Immediate Feedback with opportunities to re-submit without penalty● 3D Assessments / Tiered Assessments● Go Further Activities		
<p>Resources (hyperlink to model lessons and/or resources): Resources are created and shared within the professional learning community (PLC) of all Physical Science Teachers. We collaborate on creating quality learning experiences for all students within the classroom environment.</p> <p>NGSS Physical Science Framework: https://www.nextgenscience.org/sites/default/files/HS%20PS%20topics%20combined%206.12.13.pdf</p> <p>Discovery Education / GPB Physics in Motion Series/ CK12</p> <p>Holt Science Spectrum Physical Science Textbook</p> <p>Work and Energy Textbook Chapter (Sections 2-3) https://drive.google.com/file/d/1ileCCV7DvbY5R8OG2fTwJWtjCJkPbdG/view?usp=sharing</p> <p>Thermal Energy Textbook Chapter https://drive.google.com/file/d/1pJCRVzGStsndyiqnmKCSBjPSketghq82/view?usp=sharing</p> <p>Energy Sources and the Environment Textbook Chapter</p>			

https://drive.google.com/file/d/1Maa_beog6lOq_ymcShCXY2uY_zh6FBQU/view?usp=sharing

Group Schoology Resources

Other Sites for Interactives and Practice:

Positive Physics

The Physics Classroom

Phet Simulations

GSE Website for Physical Science

<https://www.georgiastandards.org/Georgia-Standards/Pages/Science-Physical-Science.aspx>

Reflection: Considering the planning, process and impact of the inquiry

Prior to teaching the unit	During teaching	After teaching the unit
After teaching the units as written last year, nuclear should be moved to this unit and not in the chemistry section of the course. Find engaging ways for students to explore heat transformations	Revisit previous units - spiral concepts and vocabulary that were introduced earlier Examples: Isotopes, phase change, molecular motion	Incorporate some of the optional content so students are aware of the energy resources and their impacts on the environment